# Molecular-level assessment of disease-relevant mechanisms by AFM

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### **Biomolecules at surfaces**





### Biomolecules at the solid-liquid interface

The presence of surfaces has a strong effect on biomolecules, i.e. DNA, proteins, and peptides.

#### $\rightarrow$ exploited for the self-assembly of supramolecular nanopatterns



Mamdouh *et al.*, *J. Am. Chem. Soc.* (2006), **128**, 13305



Mamdouh *et al.*, *J. Am. Chem. Soc.* (2008), **130**, 695



Bald et al., Small (2011), 7, 939

#### However -

also in the physiological environment, different surfaces interact with biomolecules: *cell membranes, blood vessels, surfaces of bones, implants, tubing, ...* 

 $\rightarrow$  How do surfaces influence the physiological action of biomolecules?

 $\rightarrow$  in-vitro studies using biological model surfaces





#### Surface properties influence protein adsorption



#### R н U A A S



## Tuning the hydrophobicity of mica surfaces by hyperthermal Ar ion irradiation





#### The structure of mica surfaces



- layered structure:
  - negatively charged aluminosilicate sheets
  - alternating layers of K<sup>+</sup> ions
  - distance between layers 1 nm
  - cleaving (Scotch tape) → atomically flat terraces → great for AFM!
- aqueous solution:
  - K<sup>+</sup> ions exchanged into solution
  - surface exhibits a negative net charge
  - $\rightarrow$  hydrophilic surface





#### Ion-beam modification of mica surfaces



#### 25 eV Ar<sup>+</sup> $\rightarrow$ mica:

- negligible effect on crystal structure
- very low sputter yield
  - $\rightarrow$  negligible effect on topography
  - $\rightarrow$  negligible effect on surface composition
- no new chemical species
- very effecient removal of outermost K<sup>+</sup> ions
- $\rightarrow$  underlying alumino-silicate sheet exposed
- $\rightarrow$  silicate tetrahedra act as adsorption sites for C
- $\rightarrow$  increased adsorption of hydrocarbons from the environment
- $\rightarrow$  hydrocarbons are hydrophobic





### Ion-induced surface hydrophobicity



### Chemical analysis by XPS



TABLE I. Elemental concentrations of K and C (in at.%) of virgin and irradiated mica surfaces aged for different times.

	Virgin			$25 \text{ eV}, 10^{15} \text{ cm}^{-2}$	
Element	10 min	4 days	64 days	4 days	67 days
K	$7.3 \pm 0.1$	$7.6 \pm 0.3$	$6.1 \pm 1.6$	$4.6 \pm 0.4$	$2.2 \pm 0.3$
С	$6.1\pm0.1$	$5.2 \pm 1.4$	$10.5\pm3.7$	$10.2\pm4.6$	$18.4\pm0.6$

- increased C content on bombarded samples
   → thin HC film
  - high fluences: preferential sputtering
- C content increasing with age
- HR XPS: composition of HC films independent of age and treatment
- contact angle not (solely) determined by amount of HCs on the surface

Keller et al., J. Chem. Phys. (2011) 134, 104705





Influence of hydrophobicity on the surface-catalyzed assembly of the Islet Amyloid Polypeptide





### Amyloid aggregation

- Polymerization of misfolded peptides/proteins in solution or at interfaces
- Self-assembly into nanostructures:
  oligomeric particles protofibrils



Dong et al., Nanotechnology. (2006), 17, 4003

higher-order fibrils



glucagon

- In principle, ANY protein or polypeptide can form amyloid aggregates!
- Amyloid aggregation related to the development of various diseases, e.g. *Alzheimer's disease*, *Parkinson's disease*, *prion disease*, ...



### The Islet Amyloid Polypeptide (IAPP)

- Islet Amyloid Polypeptide (IAPP):
  - hormonal factor secreted from the pancreatic  $\beta$ -cells together with insulin
  - reduces insulin sensitivity
  - islet amyloid deposits present the pancreas of > 90% of type II diabetes patients









### Cytotoxicity of IAPP aggregates



Engel *et al., Proc. Natl. Acad. Sci. USA.* (2008), **105**, 6033

Khemtéourian et al., Exp. Diabetes Res. (2008), 421287

#### $\rightarrow$ islet amyloid can induce apoptotic cell-death in insulin-producing $\beta$ -cells

 $\rightarrow$  relevant to the development of type II diabetes





### Amyloid aggregation at surfaces

The presence of a surface may:

- induce the formation of initial oligomers (nucleation)
- influence the assembly rate and lag time
- affect the structure of the aggregates

*Physicochemical surface properties have a strong effect on amyloid aggregation!*  Amyloid β

Mica: hydrophilic



 $\alpha$ -synuclein





Kowalewski *et al., Proc. Natl. Acad. Sci. USA*. (1999), **96**, 3688



Hoyer *et al.*, *J. Mol. Biol.* (2004), **340**, 127





### Time-lapse study of IAPP aggregation on HC films

#### Preparation of mica surfaces

- sub-100 eV ion bombardment
- exposure to lab atmosphere
- comparing aggregation on surfaces with different contact angles

#### IAPP incubation

- 13 μM IAPP in *water*
- *room temperature*
- incubation time 0.5 to 6 h
- samples dried in N<sub>2</sub> stream
- Ex-situ AFM:
  - tapping mode in air





#### IAPP on a mica/HC surface with $\theta = 23^{\circ}$



**protofibrils** height: (2.0 ± 0.5) nm

fibrils height:  $(4.5 \pm 0.5)$  nm twist:  $(28.4 \pm 3.7)$  nm





#### IAPP on a mica/HC surface with $\theta = 38^{\circ}$







#### IAPP on a mica/HC surface with $\theta = 76^{\circ}$









### Influence of hydrophobicity on IAPP aggregation



Interplay between electrostatic and hydrophobic interactions with the substrate:

 $\rightarrow$  determines conformation of adorbed monomers

 $\rightarrow$  dictates the pathway of aggregation

IAPP: 37 amino-residue polypeptide

amyloidogenic region





#### Conclusion

- Surface-catalyzed processes play an important role in the physiological environment and strongly affect how biomolecules behave.
- Disease-relevant aggregation of proteins and peptides can follow completely different pathways when occurring at surfaces.
- The physicochemical surface properties dictate the type of interaction with the molecule: *electrostatic vs. hydrophobic interactions*

 $\rightarrow$  biological model surfaces with tunable properties needed!

- Low-energy ion-beam irradiation allows the fabrication of novel model surfaces with tunable properties.
- IAPP aggregation (lag time, aggregation rate, morphology) at surfaces is driven by the interplay between electrostatic and hydrophobic interactions.





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