Introduction to electron transport in molecular systems

Reviews: Annu. Rev. Phys. Chem. 52, 681– 750 (2001) [http://atto.tau.ac.il/~nitzan/nitzanabs.html/#213] Science, 300, 1384-1389 (2003); MRS Bulletin, 29, 391-395 (2004);

Bulletin of the Israel Chemical Society, Issue 14, p. 3-13 (Dec 2003) (Hebrew)



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Molecular Rectifiers

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Received 10 June 1974 Abstract

The construction of a very simple electronic device, a rectifier, based on the use of a single organic molecule is discussed. The molecular rectifier consists of a donor pi system and an acceptor pi system, separated by a sigmabonded (methylene) tunnelling bridge. The response of such a molecule to an applied field is calculated, and rectifier properties indeed appear.

<u>Xe on Ni(110)</u>



Moore's "Law"

Scaling of electronic devices



Moore's 2nd law



Molecules get wired





Antibiotic-resistant enterococci Islands of pathogenicity Asteroid families Aftermath of a recent break-up Non-human primates When can experiments be justified?.

new on the market microscopy



IEEE TRANSACTIONS ON ELECTRON DEVICES VOL.43 OCTOBER **1996** 1637

Need for Critical Assessment Rolf Landauer,*Life Fellow,IEEE* Abstract

Adventurous technological proposals are subject to inadequate critical assessment. It is the proponents who organize meetings and special issues. Optical logic, mesoscopic switching devices and quantum parallelism are used to illustrate this problem.

This editorial, disguised as a scientific paper, is obviously a plan for more honesty. We do not, in the long run, build effective public support for science and technology by promising more than we can deliver.

Feynman

• For a successful Technology, reality must take precedence over public relations, for nature cannot be fooled

First Transport Measurements through Single Molecules

Adsorbed molecule addressed by STM tip

Self-assembled monolayers



Dorogi *et al.* PRB 52 (95) @ Purdue

Molecule between two electrodes

Break junction: dithiols between gold



Reed et al. Science 278 (97) @ Yale

Molecule lying on a surface

Single-wall carbon nanotube on Pt



Dekker et al. Nature 386(97)

<u>C₆₀ on gold</u>



Joachim et al. PRL 74 (95)

Nanopore



Reed et al. APL 71 (97)



Lieber et al. Nature 391 (98)



Park et. al. Nature 417,722-725 (2002)





Weber et al, Chem. Phys. 2002









Xu et al (Tao), NanoLet (2004)

loge of GCGC(AT)mGCGC conductance vs length (total number of base pairs). The solid line is a linear fit that reflects the exponential dependence of the conductance on length. The decay constant, , is determined from the slope of the linear fit. (b) Conductance of (GC)n vs 1/length (in total base pairs).

AT THE BEGINNING...



Activated rate processes



The physics of transition state rates

Assume:

 $-\infty$

(1) Equilibrium in the well



(2) Every trajectory on the barrier that goes out makes it

$$k_{TST} = \int_{0}^{\infty} dv \, v \, P(x_B, v) = \langle v_f \rangle P(x_B) = \frac{\omega_0}{2\pi} e^{-\beta E_B}$$

$$\int_{0}^{\infty} dv v e^{-\frac{1}{2}\beta m v^2} = \frac{1}{\sqrt{2\pi\beta m}}$$

$$P(x_B) = \frac{\exp(-\beta E_B)}{\int_{-\infty}^{E_B} dx \exp(-\beta V(x))} = \sqrt{\frac{\beta m \omega_0^2}{2\pi}} e^{-\beta E_B}$$

Theory of Electron Transfer

Activation energy
Transition probability
Rate – Transition state theory or solvent controlled

Electron transfer in polar media

•Electron are much faster than nuclei

• Electronic transitions take place in fixed nuclear configurations

• Electronic energy needs to be conserved during the change in electronic charge density





Electron transition takes place in unstable nuclear configurations obtained via thermal fluctuations

Electron transfer



Transition
Alternatively –
solvent control
Adiabatic and non-adiabatic ET processes



Electron transfer: Marcus theory



Electron transfer: Effect of Driving (=energy gap)



Experimental confirmation of the inverted regime



Miller et al, JACS(1984)

Marcus Nobel Prize: 1992

Electron transfer – the coupling

From Quantum Chemical Calculations
The Mulliken-Hush formula

$$\mid H_{DA} \mid = \frac{\hbar \omega_{\max} \left| \mu_{12} \right|}{e R_{DA}}$$

Bridge mediated electron transfer

$$\hat{V}_{12,7} = \frac{V_{23}}{2} + \frac{V_{23}}{3} + \frac{V_{3A}}{3} + \frac{V_{$$

Marcus expresions for non-adiabatic ET rates

$$k_{D \to A} = \frac{2\pi}{\hbar} |V_{DA}|^2 \mathcal{F}(E_{AD})$$

$$= \frac{2\pi}{\hbar} |V_{D1}V_{NA}|^2 |G_{1N}^{(B)}(E_D)|^2 \mathcal{F}(E_{AD})$$

Donor-to-Bridge/
Acceptor-to-bridge
Bridge Green's
Function

$$\mathcal{F}(E) = \frac{e^{-(\lambda + E)^2/4\lambda k_B T}}{\sqrt{4\pi\lambda k_B T}}$$
Franck-Condon-
weighted DOS
Reorganization energy

Bridge mediated ET rate

$k_{ET} \sim \mathcal{F}(E_{AD}, T) \exp(-\beta' R_{DA})$

β' (Å⁻¹)=
0.2-0.6 for highly conjugated chains
0.9-1.2 for saturated hydrocarbons
~2 for vacuum

Incoherent hopping



ET rate from steady state hopping

$0 = -(k_{0,1} + k_{2,1})P_1 + k_{1,0}P_0 + k_{1,2}P_2$

$$0 = -(k_{N-1,N} + k_{N+1,N})P_N + k_{N,N-1}P_{N-1}$$
$$\dot{P}_{N+1} = k_{N+1,N}P_N = k_{D\to A}P_0$$

$$k_{D \to A} \equiv k_{N+1,0} = \frac{k e^{-(E_B / k_B T)}}{\left[\frac{k}{k_{N \to A}} + \frac{k}{k_{1 \to D}} - 1\right] + N}$$

Dependence on temperature



The integrated elastic (dotted line) and activated (dashed line) components of the transmission, and the total transmission probability (full line) displayed as function of inverse temperature. Parameters are as in Fig. 3.

The photosythetic reaction center



Michel - Beyerle et al



DNA (Giese et al 2001)



Steady state evaluation of rates

h

Rate of water flow depends linearly on water height in the cylinder

Two ways to get the rate of water flowing out:

(1)Measure h(t) and get the rate
 coefficient from k=(1/h)dh/dt

(2) Keep h constant and measure the steady state outwards water flux J. Get the rate from k=J/h -- Steady state rate

Steady state quantum mechanics



Starting from state 0 at t=0: $P_0 = \exp(-\Gamma_0 t)$ $\Gamma_0 = 2\pi |V_{0l}|^2 \rho_L$ (Golden Rule)

•

Steady state derivation:

 $\psi(t) = \overline{C_0(t)} |0\rangle + \sum_l \overline{C_l(t)} |l\rangle$

$$C_{0} \stackrel{\neq}{=} \stackrel{a}{\underbrace{c_{0}}} \underbrace{\overline{c_{0}}}_{l} \stackrel{=}{=} \stackrel{i}{=} \underbrace{\underline{F_{0}}}_{0} \underbrace{c_{0}}_{l} \stackrel{=}{=} \underbrace{\overline{F_{0}}}_{l} \underbrace{F_{0}}_{l} \stackrel{=}{=} \underbrace{\overline{F_{0}}}_{l} \underbrace{F_{0}}_{l} \stackrel{=}{=} \underbrace{\overline{F_{0}}}_{l} \underbrace{F_{0}}_{l} \stackrel{=}{=} \underbrace{F_{0}}}_{l} \stackrel{=}{=} \underbrace{F_{0}}}_{l} \underbrace{F_{0}}_{l} \stackrel{=}{=} \underbrace{F_{0}}_{l} \underbrace{F_{0}}_{l} \stackrel{=}{=} \underbrace{F_{0}}_{l} \underbrace{F_{0}}_{l} \stackrel{=}{=} \underbrace{F_{0}}}_{l} \underbrace{F_{0}}_{l} \stackrel{=}{=} \underbrace{F_{0}}_{l} \underbrace{F_{0}}_{l} \stackrel{=}{=} \underbrace{F_{0}}_{l} \stackrel{=}{=} \underbrace{F_{0}}_{l} \underbrace{F_{0}}_{l} \stackrel{=}{=} \underbrace{F_{0}}_{l} \underbrace{F_{0}}_{l} \stackrel{=}{=} \underbrace{F_{0}}_{l} \underbrace{F_{0}}_{l} \stackrel{=}{=}$$

$$C_{0} = c_{0}e^{-(i/\hbar)E_{0}t} \quad \text{pumping} \quad \text{damping}$$

$$\hbar \frac{d}{dt}C_{l} = -iE_{l}C_{l} - (iV_{l0}C_{0}) - (1/2)\eta C_{l} ; \text{ all } l$$

$$C_{l}(t) = c_{l}e^{-(i/\hbar)E_{0}t} ; \quad c_{l} = \frac{V_{ls}c_{0}}{E_{0} - E_{l} + i\eta/2}$$

$$J = (\eta/\hbar)\sum_{I}|C_{l}|^{2} = |C_{0}|^{2}\sum_{I}|V_{l0}|^{2} \frac{\eta/\hbar}{(E_{0} - E_{l})^{2} + (\eta/2)^{2}}$$

$$-\frac{\eta \rightarrow 0}{\rho}|C_{0}|^{2} \frac{2\pi}{\hbar}\sum_{I}|V_{l0}|^{2} \delta(E_{0} - E_{l})$$

$$k = \frac{J}{|C_{0}|^{2}} = \frac{2\pi}{\hbar}\sum_{I}|V_{l0}|^{2} \delta(E_{0} - E_{l}) = \frac{2\pi}{\hbar}(|V_{l0}|^{2}\rho_{L})_{E_{l}=E_{0}} = \Gamma_{0}/\hbar$$

Resonance scattering



 $V = V_{0,1} \left| 0 \right\rangle \left\langle 1 \right| + V_{1,0} \left| 1 \right\rangle \left\langle 0 \right| + \sum_{l} \left(V_{l,1} \left| l \right\rangle \left\langle 1 \right| + V_{1,l} \left| 1 \right\rangle \left\langle l \right| \right) + \sum_{r} \left(V_{r,1} \left| r \right\rangle \left\langle 1 \right| + V_{1,r} \left| 1 \right\rangle \left\langle r \right| \right)$

Resonant tunneling



$$J_{0 \to R} = \frac{|V_{1,0}|^2}{\left(E_0 - E_1\right)^2 + \left(\Gamma_1 / 2\right)^2} \frac{\Gamma_{1R}}{\hbar} |c_0|^2$$



Resonant Tunneling



Resonant Transmission – 3d



$$\mathcal{T}(E_0) = \frac{\Gamma_{1L}(E_0)\Gamma_{1R}(E_0)}{\left(E_0 - \tilde{E}_1\right)^2 + \left(\Gamma_1(E_0)/2\right)^2}$$
$$\Gamma_1 = \Gamma_{1R} + \Gamma_{1L} \qquad 1d$$

3d: Total flux from L to R at energy \mathbf{E}_0 : $\left(\frac{dJ_{L\to R}(E)}{dE}\right)_{E=E_0} = \frac{1}{\left(2\pi\hbar\right)} \frac{\Gamma_{1L}(E_0)\Gamma_{1R}(E_0)}{\left(E_0 - \tilde{E}_1\right)^2 + \left(\Gamma_1(E_0)/2\right)^2} |c_0|^2$

If the continua are associated with a metal electrode at thermal equilibrium than $|c_0|^2 = f(E_0) = \left[\exp\left((E_0 - \mu)/k_BT\right) + 1\right]^{-1}$ (Fermi-Dirac distribution)



For a single "channel":

 $\mathcal{T}(E) = \frac{\Gamma_{1L}(E)\Gamma_{1R}(E)}{\left(E - \tilde{E}_1\right)^2 + \left(\Gamma_1(E)/2\right)^2} \quad (\text{maximum=1})$

<u>()</u>

Maximum conductance per channel

Molecular level structure between electrodes











Cui et al (Lindsay), Science 294, 571 (2001)

"The resistance of a single octanedithiol molecule was 900±50 megaohms, based on measurements on more than 1000 single molecules. In contrast, nonbonded contacts to octanethiol monolayers were at least four orders of magnitude more resistive, less reproducible, and had a different voltage dependence, demonstrating that the measurement of intrinsic molecular properties requires chemically bonded contacts".

-> Wide band approximation



ET vs Conduction



$$g = \frac{e^2}{\pi\hbar} |G_{0,N+1}(E)|^2 \Gamma_0^{(L)}(E) \Gamma_{N+1}^{(R)}(E)$$

$$= \frac{e^2}{\pi\hbar} \frac{V_{01}V_{N,N+1}}{\left(E - E_D + \frac{1}{2}i\Gamma_0^{(L)}\right) \left(E - E_A + \frac{1}{2}i\Gamma_{N+1}^{(R)}\right)}^2 |G_{1N}^{(B)}(E)|^2 \Gamma_0^{(L)}(E) \Gamma_{N+1}^{(R)}(E)$$

$$k_{D \to A} = \frac{2\pi}{\hbar} |V_{DA}|^2 \mathcal{F}(E_{AD})$$

$$= \frac{2\pi}{\hbar} |V_{01}V_{N,N+1}|^2 |G_{1N}^{(B)}(E_D)|^2 \mathcal{F}(E_{AD})$$

A relation between g and k



A relation between g and k

$$g \approx \frac{8e^2}{\pi^2 \Gamma_D^{(L)} \Gamma_A^{(R)} \mathcal{F}} k_{D \to A}$$

$$\mathcal{F} = \left(\sqrt{4\pi\lambda k_B T}\right)^{-1} \exp\left(-\lambda/4k_B T\right)$$
$$\lambda \approx 0.5 \text{eV} \qquad \Gamma_D^{(L)} = \Gamma_A^{(R)} \approx 0.5 \text{eV}$$

$$g \sim \left(e^{2} / \pi \hbar \right) \left(10^{-13} k_{D \to A} (s^{-1}) \right)$$
$$\cong \left[10^{-17} k_{D \to A} (s^{-1}) \right] \Omega^{-1}$$

Comparing conduction to rates (M. Newton, 2003)



2-level bridge (local representation)



•Molecule-electrode coupling Γ_L , Γ_R •Molecular energetics E_1 , E_2 •Intramolecular coupling $V_{1,2}$

Spin



Ratner and Troisi, 2004



"Switching"



Reasons for switching

Conformational changes

Transient charging



STM under water S.Boussaad et. al. JCP (2003)



Tsai et. al. PRL 1992: RTS in Me-SiO₂-Si junctions

Temperature and chain length dependence



Where does the potential bias falls, and how?

•Image effect

•Electron-electron interaction (on the Hartree level)









Experiment

Theoretical Model



Experimental i/V behavior



Potential distribution



NEGF - HF calculation



$HS - CH_2CH_2CH_2CH_2CH_2CH_3 - CH_3CH_2 - SH$

